

$wR = 0.047$
 $S = 1.49$
 906 reflections
 137 parameters
 H atoms were refined with one overall U_{iso}

$(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.17 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.27 \text{ e } \text{\AA}^{-3}$
 Extinction correction: none
 Atomic scattering factors from *SHELX76* (Sheldrick, 1976)

Lists of structure factors, anisotropic displacement parameters and H-atom coordinates have been deposited with the IUCr (Reference: LI1118). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | B_{eq} |
|-------|------------|-------------|------------|-----------------|
| C1 | 0.6410 (1) | 0.0435 (3) | 0.5583 (1) | 6.04 (5) |
| N | 1.0358 (3) | -0.0072 (6) | 0.6319 (2) | 3.3 (1) |
| O(1) | 0.9848 (3) | 0.3551 (6) | 0.7022 (2) | 4.4 (1) |
| O(2) | 1.2610 (3) | 0.2545 (7) | 0.7665 (2) | 5.5 (1) |
| O(3) | 0.9000 (3) | -0.2222 (7) | 0.5490 (2) | 5.9 (1) |
| C(2) | 1.0612 (4) | 0.1833 (9) | 0.6855 (2) | 3.6 (1) |
| C(3) | 1.2090 (4) | 0.1287 (9) | 0.7174 (2) | 4.0 (1) |
| C(4) | 1.3891 (5) | -0.206 (1) | 0.6820 (2) | 4.7 (2) |
| C(5) | 1.4137 (5) | -0.410 (1) | 0.6353 (3) | 5.0 (2) |
| C(6) | 1.3092 (5) | -0.4888 (9) | 0.5857 (2) | 4.8 (2) |
| C(7) | 1.1803 (4) | -0.3712 (8) | 0.5802 (2) | 4.1 (2) |
| C(8) | 1.1576 (4) | -0.1680 (8) | 0.6270 (2) | 3.5 (1) |
| C(9) | 1.2607 (4) | -0.0892 (8) | 0.6768 (2) | 3.7 (1) |
| C(10) | 0.9112 (4) | -0.0469 (9) | 0.5917 (2) | 3.8 (1) |
| C(11) | 0.7953 (4) | 0.1402 (9) | 0.6066 (2) | 4.2 (1) |

Acta Cryst. (1995). **C51**, 676–678

7,10-Dimethoxybenzo[*b*]carbazole†

J. SEETHARAMAN AND S. S. RAJAN*

Department of Crystallography and Biophysics,
 University of Madras, Madras 600025, India

(Received 10 November 1993; accepted 12 May 1994)

Abstract

The asymmetric unit of the title compound, $C_{18}H_{15}NO_2$, contains two crystallographically independent molecules with similar conformations. Each molecule consists of a carbazole ring system and a fused substituted benzene ring. The molecules are highly planar and their conformations are similar to that of ellipticine [Courseille, Busetta & Hospital (1974). *Acta Cryst.* **B30**, 2628–2631], which is a DNA-intercalating molecule.

Comment

Preliminary studies of carbazole derivatives have shown that the presence of oxygenated substituents increases their biological activity (Hewlins, Oliveira-Campos & Shannon, 1984). Their structures are analogous to that of ellipticine (Courseille, Busetta & Hospital, 1974), a plant alkaloid having pronounced anti-tumour activity (Hartwell & Abbott, 1969), and they have been found to have DNA-intercalating properties (Neidle, 1979; Gale, Cundliffe, Reynolds,

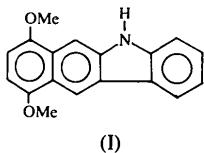
Table 2. Selected geometric parameters (\AA , °)

| | | | |
|----------------|-----------|------------------|-----------|
| C1—C(11) | 1.768 (4) | N—C(2) | 1.408 (5) |
| N—C(8) | 1.442 (5) | N—C(10) | 1.398 (5) |
| O(1)—C(2) | 1.196 (5) | O(2)—C(3) | 1.209 (5) |
| O(3)—C(10) | 1.198 (5) | C(2)—C(3) | 1.542 (6) |
| C(3)—C(9) | 1.452 (6) | C(4)—C(5) | 1.390 (7) |
| C(4)—C(9) | 1.376 (6) | C(5)—C(6) | 1.393 (7) |
| C(6)—C(7) | 1.382 (6) | C(7)—C(8) | 1.385 (6) |
| C(8)—C(9) | 1.386 (6) | C(10)—C(11) | 1.510 (6) |
| C(2)—N—C(8) | 109.1 (3) | C(6)—C(7)—C(8) | 117.0 (4) |
| C(2)—N—C(10) | 126.3 (3) | N—C(8)—C(7) | 128.7 (4) |
| C(8)—N—C(10) | 124.5 (3) | N—C(8)—C(9) | 110.5 (3) |
| N—C(2)—O(1) | 127.2 (4) | C(7)—C(8)—C(9) | 120.7 (4) |
| N—C(2)—C(3) | 105.9 (3) | C(3)—C(9)—C(4) | 129.1 (4) |
| O(1)—C(2)—C(3) | 126.9 (4) | C(3)—C(9)—C(8) | 108.6 (4) |
| O(2)—C(3)—C(2) | 122.4 (4) | C(4)—C(9)—C(8) | 122.2 (4) |
| O(2)—C(3)—C(9) | 131.8 (4) | N—C(10)—O(3) | 120.5 (4) |
| C(2)—C(3)—C(9) | 105.8 (3) | N—C(10)—C(11) | 115.8 (3) |
| C(5)—C(4)—C(9) | 117.8 (4) | O(3)—C(10)—C(11) | 123.6 (4) |
| C(4)—C(5)—C(6) | 119.6 (4) | C1—C(11)—C(10) | 110.4 (3) |
| C(5)—C(6)—C(7) | 122.7 (4) | | |

Data were corrected for Lorentz, polarization and absorption effects. The structure was solved by direct methods. H atoms were found in a difference synthesis and included as fixed contributors with an overall isotropic displacement factor that refined to $U_{\text{iso}} = 0.081 (6) \text{ \AA}^2$. Programs used were: *SHELXS86* (Sheldrick, 1985), *SHELX76* (Sheldrick, 1976) and *ORTEP* (Johnson, 1965). Most of the calculations were performed on a MicroVAX 3600 computer at the Weizmann Institute of Science, Israel.

This work has received partial support from FAPESP, CNPq, CAPES and FINEP. One of us (JZ-S) would like to thank the Associação de Amigos do Instituto Weizmann de São Paulo for a scholarship.

Richmond & Waring, 1981; Aggarwal, Neidle & Sainsbury, 1983). Ellipticine derivatives with various substituents have been synthesized and their DNA affinity and anti-tumour activities have been tested (Le Pecq, Xuong, Gosse & Paoletti, 1974; Paoletti, Cros, Xuong, Lecomte & Moisand, 1979). The title compound, (I), a carbazole derivative similar to ellipticine, was synthesized by means of the reductive condensation method (Rajeswaran & Srinivasan, 1994) and was crystallized from acetone.



(I)

A perspective view (*PLUTO*; Motherwell & Clegg, 1976) of the molecule is shown in Fig. 1. The mean values of the $C(sp^2)-C(sp^2)$ bond lengths in molecules *A* and *B* are 1.397 (2) and 1.397 (1) Å, respectively, and they agree well with those in related compounds (Aggarwal, Neidle & Sainsbury, 1983; Reimers, Guth & Wang, 1984). The molecules are highly planar [$\sum(\Delta/\sigma)^2 = 32$ and 39 for molecules *A* and *B*, respectively], as found in ellipticine derivatives such as 5-*n*-butyl-11-dimethylellipticine and 9-methoxyellipticine (Aggarwal, Neidle & Sainsbury, 1983), 7-methylellipticine (Kuroda & Sainsbury, 1984) and ellipticine-iodoCpG⁺ (Jain, Bhandary & Sobell, 1979). This planarity is said to be essential for intercalation with DNA (Neidle, 1979; Gale,

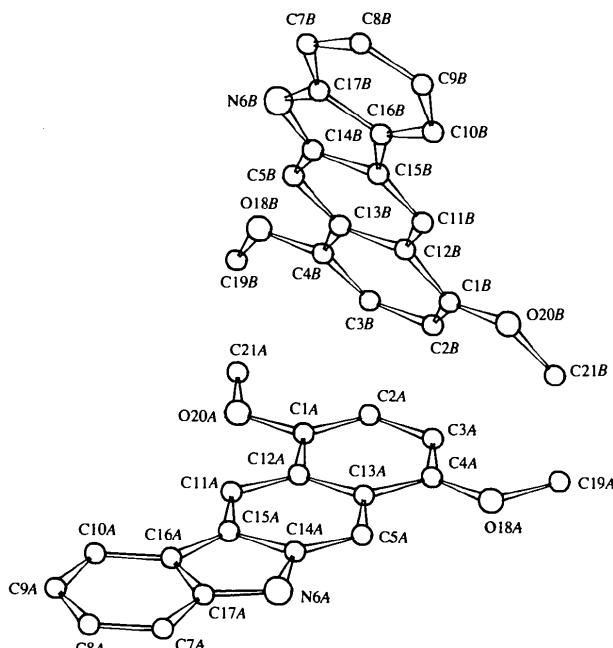


Fig. 1. Perspective view of the molecules with the atomic numbering scheme.

Cundliffe, Reynolds, Richmond & Waring, 1981; Aggarwal, Neidle & Sainsbury, 1983). The deviations of the methoxy C atoms from the plane of the benzene ring are 0.038 (9) and 0.050 (10) Å in molecule *A* and 0.194 (8) and 0.109 (10) Å in molecule *B*. The molecules in the unit cell are held together by van der Waal forces.

Experimental

Crystal data

| | |
|---------------------------------|---|
| $C_{18}H_{15}NO_2$ | $Cu K\alpha$ radiation |
| $M_r = 277.31$ | $\lambda = 1.5418 \text{ \AA}$ |
| Trigonal | Cell parameters from 20 reflections |
| $P\bar{3}_1$ | $\theta = 20-30^\circ$ |
| $a = 12.294 (3) \text{ \AA}$ | $\mu = 0.62 \text{ mm}^{-1}$ |
| $c = 15.945 (3) \text{ \AA}$ | $T = 293 \text{ K}$ |
| $V = 2087 (3) \text{ \AA}^3$ | Rectangular |
| $Z = 6$ | $0.35 \times 0.25 \times 0.20 \text{ mm}$ |
| $D_x = 1.324 \text{ Mg m}^{-3}$ | Colourless |

Data collection

| | |
|---------------------------------------|----------------------------|
| Enraf–Nonius CAD-4 | $\theta_{\max} = 70^\circ$ |
| diffractometer | $h = -14 \rightarrow 14$ |
| $\omega-2\theta$ scans | $k = 0 \rightarrow 13$ |
| Absorption correction: | $l = 0 \rightarrow 19$ |
| empirical | 3 standard reflections |
| $T_{\min} = 0.84$, $T_{\max} = 0.97$ | monitored every 100 |
| 2258 measured reflections | reflections |
| 2258 independent reflections | intensity decay: <2% |
| 1525 observed reflections | |
| $[I > 3\sigma(I)]$ | |

Refinement

| | |
|-------------------------------|--|
| Refinement on F | $w = 1/[\sigma^2(F) + 0.00259F^2]$ |
| $R = 0.043$ | $(\Delta/\sigma)_{\max} = 0.041$ |
| $wR = 0.046$ | $\Delta\rho_{\max} = 0.30 \text{ e \AA}^{-3}$ |
| $S = 0.97$ | $\Delta\rho_{\min} = -0.21 \text{ e \AA}^{-3}$ |
| 1525 reflections | Extinction correction: none |
| 499 parameters | Atomic scattering factors from <i>SHELX76</i> |
| All H-atom parameters refined | (Sheldrick, 1976) |

Table 1. Fractional atomic coordinates and equivalent isotropic displacement parameters (\AA^2)

| | x | y | z | U_{eq} |
|------|------------|------------|-------------|-----------|
| C1A | 0.6973 (5) | 0.3160 (6) | -0.0946 (4) | 0.051 (5) |
| C2A | 0.6500 (6) | 0.3910 (7) | -0.0710 (5) | 0.061 (6) |
| C3A | 0.7057 (7) | 0.5143 (7) | -0.1015 (5) | 0.074 (7) |
| C4A | 0.8069 (6) | 0.5633 (6) | -0.1521 (4) | 0.052 (5) |
| C5A | 0.9675 (5) | 0.5377 (5) | -0.2284 (4) | 0.043 (5) |
| N6A | 1.1167 (5) | 0.4863 (5) | -0.3012 (4) | 0.055 (5) |
| C7A | 1.2182 (6) | 0.3635 (7) | -0.3500 (5) | 0.069 (7) |
| C8A | 1.2112 (7) | 0.2498 (8) | -0.3441 (5) | 0.079 (8) |
| C9A | 1.1174 (7) | 0.1528 (7) | -0.2966 (6) | 0.080 (7) |
| C10A | 1.0292 (6) | 0.1687 (6) | -0.2520 (5) | 0.063 (6) |
| C11A | 0.8551 (5) | 0.2863 (5) | -0.1713 (4) | 0.046 (5) |
| C12A | 0.8039 (5) | 0.3615 (6) | -0.1487 (4) | 0.046 (5) |
| C13A | 0.8620 (5) | 0.4884 (6) | -0.1764 (4) | 0.048 (5) |
| C14A | 1.0131 (5) | 0.4601 (5) | -0.2500 (4) | 0.045 (5) |

| | | | | |
|------|------------|------------|-------------|-----------|
| C15A | 0.9589 (5) | 0.3335 (5) | -0.2228 (4) | 0.046 (5) |
| C16A | 1.0342 (6) | 0.2846 (6) | -0.2580 (4) | 0.051 (5) |
| C17A | 1.1288 (6) | 0.3802 (6) | -0.3057 (4) | 0.053 (5) |
| O18A | 0.8688 (4) | 0.6830 (4) | -0.1833 (3) | 0.071 (4) |
| C19A | 0.8227 (7) | 0.7627 (7) | -0.1584 (6) | 0.090 (8) |
| O20A | 0.6516 (4) | 0.1940 (4) | -0.0680 (3) | 0.065 (4) |
| C21A | 0.5447 (7) | 0.1404 (8) | -0.0142 (6) | 0.080 (7) |
| C1B | 1.1408 (5) | 0.7573 (5) | 0.0314 (4) | 0.048 (5) |
| C2B | 1.0217 (6) | 0.6721 (6) | 0.0118 (5) | 0.063 (5) |
| C3B | 0.9648 (6) | 0.5486 (5) | 0.0437 (5) | 0.052 (5) |
| C4B | 1.0305 (5) | 0.5118 (5) | 0.0945 (4) | 0.043 (4) |
| C5B | 1.2242 (5) | 0.5650 (5) | 0.1731 (4) | 0.046 (5) |
| N6B | 1.4302 (5) | 0.6457 (4) | 0.2470 (4) | 0.054 (4) |
| C7B | 1.6466 (6) | 0.7979 (6) | 0.3003 (5) | 0.057 (5) |
| C8B | 1.7390 (6) | 0.9234 (7) | 0.2955 (5) | 0.065 (6) |
| C9B | 1.7260 (6) | 1.0098 (6) | 0.2483 (5) | 0.064 (6) |
| C10B | 1.6211 (5) | 0.9730 (5) | 0.1997 (4) | 0.050 (5) |
| C11B | 1.3360 (5) | 0.8145 (5) | 0.1122 (4) | 0.045 (4) |
| C12B | 1.2125 (5) | 0.7260 (5) | 0.0882 (4) | 0.044 (5) |
| C13B | 1.1558 (5) | 0.6000 (5) | 0.1195 (4) | 0.043 (4) |
| C14B | 1.3461 (6) | 0.6551 (5) | 0.1954 (4) | 0.052 (5) |
| C15B | 1.4001 (5) | 0.7801 (5) | 0.1647 (4) | 0.042 (4) |
| C16B | 1.5246 (5) | 0.8468 (5) | 0.2026 (4) | 0.045 (4) |
| C17B | 1.5388 (5) | 0.7609 (5) | 0.2525 (4) | 0.045 (4) |
| O18B | 0.9837 (4) | 0.3928 (4) | 0.1263 (3) | 0.051 (3) |
| C19B | 0.8640 (6) | 0.2988 (6) | 0.0969 (5) | 0.062 (6) |
| O20B | 1.2049 (4) | 0.8773 (4) | 0.0020 (3) | 0.060 (3) |
| C21B | 1.1417 (7) | 0.9157 (7) | -0.0542 (6) | 0.077 (7) |

Table 2. Selected geometric parameters (\AA , $^\circ$)

- C1A—O20A 1.379 (8) C1B—O20B 1.362 (7)
 C4A—O18A 1.368 (8) C4B—O18B 1.374 (7)
 N6A—C14A 1.408 (9) N6B—C14B 1.371 (10)
 N6A—C17A 1.387 (11) N6B—C17B 1.380 (6)
 O18A—C19A 1.412 (12) O18B—C19B 1.421 (7)
 O20A—C21A 1.425 (9) O20B—C21B 1.421 (11)
 C12A—C1A—O20A 114.0 (6) C12B—C1B—O20B 113.9 (5)
 C2A—C1A—O20A 124.7 (6) C2B—C1B—O20B 125.4 (6)
 C3A—C4A—O18A 126.3 (7) C3B—C4B—O18B 124.6 (6)
 C13A—C4A—O18A 113.7 (6) C13B—C4B—O18B 115.1 (6)
 C14A—N6A—C17A 109.0 (5) C14B—N6B—C17B 109.6 (5)
 C5A—C14A—N6A 128.7 (5) C5B—C14B—N6B 130.0 (5)
 N6A—C14A—C15A 107.6 (5) N6B—C14B—C15B 109.5 (6)
 N6A—C17A—C16A 109.1 (6) N6B—C17B—C16B 108.4 (5)
 N6A—C17A—C7A 128.5 (6) N6B—C17B—C7B 130.1 (6)
 C4A—O18A—C19A 116.3 (6) C4B—O18B—C19B 117.1 (5)
 C1A—O20A—C21A 117.9 (6) C1B—O20B—C21B 117.6 (5)
- Courseille, C., Busetta, B. & Hospital, M. (1974). *Acta Cryst.* **B30**, 2628–2631.
 Frenz, B. A. (1978). *The Enraf-Nonius CAD-4 SDP – a Real-Time System for Concurrent X-ray Data Collection and Crystal Structure Solution. Computing in Crystallography*, edited by H. Schenk, R. Olthof-Hazekamp, H. van Koningsveld & G. C. Bassi, pp. 64–71. Delft Univ. Press.
 Gale, E. F., Cundliffe, E., Reynolds, P. E., Richmond, M. H. & Waring, M. J. (1981). *The Molecular Basis of Antibiotic Action*. London: John Wiley.
 Hartelli, J. R. & Abbott, B. (1969). *Adv. Pharm. Chemother.* **7**, 117–209.
 Hewlins, M. J. E., Oliveira-Campos, A. M. & Shannon, P. V. R. (1984). *Synthesis*, pp. 289–302.
 Jain, S. C., Bhandary, K. K. & Sobell, H. M. (1979). *J. Mol. Biol.* **135**, 813–840.
 Kuroda, R. & Sainsbury, M. (1984). *J. Chem. Soc. Perkin Trans. 2*, pp. 1751–1753.
 Le Pecq, J. B., Xuong, N.-D., Gosse, C. & Paoletti, C. (1974). *Proc. Natl Acad. Sci. USA*, **71**, 5078–5082.
 Motherwell, W. D. S. & Clegg, W. (1976). *PLUTO. Program for Plotting Molecular and Crystal Structures*. Univ. of Cambridge, England.
 Nardelli, M. (1983). *Comput. Chem.* **7**, 95–98.
 Neidle, S. (1979). *Prog. Med. Chem.* **16**, 151–221.
 Paoletti, C., Cros, S., Xuong, N.-D., Lecointe, P. & Moisand, A. (1979). *Chem. Biol. Interact.* **25**, 45–58.
 Rajeswaran, W. R. & Srinivasan, P. C. (1994). *Ind. J. Chem.* **33B**, 368–369.
 Reimers, W., Guth, H. & Wang, Z.-T., (1984). *Acta Cryst.* **C40**, 977–978.
 Sheldrick, G. M. (1976). *SHELX76. Program for Crystal Structure Determination*. Univ. of Cambridge, England.
 Sheldrick, G. M. (1985). *SHELXS86. Program for the Solution of Crystal Structures*. Univ. of Göttingen, Germany.

Acta Cryst. (1995). **C51**, 678–680

Absolute Configuration of *N,N*-(1*S*)-1-[(1-Methoxy-1-methyl)ethyl]tetramethylene- (1*R*,6*R*,7*R*)-9,9-diethoxy-1-methyl-2-oxo- bicyclo[4.3.0]nonane-7-sulfonamide

BERNARD TINANT AND JEAN-PAUL DECLERCQ

Laboratoire de Chimie Physique et de Cristallographie,
 Université Catholique de Louvain, 1 place Louis
 Pasteur, 1348 Louvain-la-Neuve, Belgium

CATHERINE HUART

Laboratoire de Chimie Organique de Synthèse,
 Université Catholique de Louvain, 1 place Louis
 Pasteur, 1348 Louvain-la-Neuve, Belgium

(Received 8 October 1993; accepted 8 February 1994)

Abstract

The title compound, $C_{22}H_{39}NO_6S$, is obtained by an intramolecular cyclization of the adduct resulting

Lists of structure factors, anisotropic displacement parameters, H-atom coordinates, bond distances involving H atoms, least-squares-planes data and torsion angles have been deposited with the IUCr (Reference: VJ1003). Copies may be obtained through The Managing Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

References

Aggarwal, A., Neidle, S. & Sainsbury, M. (1983). *Acta Cryst.* **C39**, 631–633.